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Research Article

## Isolation and Characterization of Bioactive Components Derived from Whole Plant of *Selaginella bryopteris*

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### ABSTRACT

Natural products and herbal remedies used in traditional folklore medicine have been the source of many medically beneficial drugs because they elicit fewer side effects, relatively cheap, affordable and claimed to be effective. However, in order to make these remedies acceptable to modern medicine, there is a need to scientifically evaluate them to identify the active principles and to understand their mechanism of action. *Selaginella bryopteris* (*S. bryopteris*) is a pteridophytic plant belongs to the family selaginellaceae. Its familiar name is sanjeevani booti. The aim of the present investigation was isolation and characterization of active components derived from whole plant of *S. bryopteris*. The plant was extracted with petroleum ether, ethyl acetate and methanol solvent. The preliminary phytochemical results revealed that alkaloids, carbohydrates, reducing sugar, flavonoids, protein, amino acid, tannin and phenolic compound as active constituents in methanolic extract of *S. bryopteris*. The total phenolics content of whole plant of methanolic extract was (184.13±0.416 mg/gm), followed by flavonoids (89.67±2.516mg/gm). The quantification and the identification of compounds in the crude extract and active bands isolated by preparative TLC were accomplished using spectroscopic analysis. The most important compounds amentoflavone identified in the crude extract appreciable amounts may account for its various biological activities.

**Keywords:** *Selaginella bryopteris*, Isolation, Plant extraction, Phytochemical

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### INTRODUCTION

The use of plants and their constituents in primary health care has ancient history as old as human beings. Various medicinal plants have proven therapeutics implication in the health management via antioxidant, anti-inflammatory, anti-diabetic, and other biological activities<sup>1,2</sup>. *S. bryopteris* is a perennial, herbaceous lythophytic plant that grows in shallow soils on rocky outcrops of slopes of small hills in direct sunlight in humid tropical regions. It is a traditional herb that for centuries has occupied a prime place among the most sought-after herbs in Indian mythology as 'Sanjeevani' (one that infuses life) by virtue of its resurrection properties. The herb is popular among tribal people of India as a dietary supplement in treatment, signifying its role as a 'panacea' against varied maladies<sup>3,4</sup>. Several studies have explored the bioactive components contributing to the medicinal properties of varied species of *Selaginella*<sup>5-10</sup>. In particular, the flavonoid-rich contents of this herb have demonstrated numerous and varied

biological activities<sup>11-14</sup>. In lieu of impressive epidemiological evidence for the cytoprotective effects of plant flavonoids and correlation of high flavonoid intake with a decreased risk of cancer. Amentoflavone (C<sub>30</sub>H<sub>18</sub>O<sub>10</sub>) is a common biflavonoid chemically named as 8-[5-(5,7-dihydroxy-4-oxo-4H-chromen-2-yl)-2-hydroxyphenyl]-5,7-dihydroxy-2-(4-hydroxyphenyl)-4H-chromen-4-one, which naturally occurs in many plants. It is also considered as an apigenin dimer linked by a C<sub>3</sub>-C<sub>8</sub> covalent bond. This compound was firstly isolated by Okigawa and his colleagues in 1971 from three plants of the *Selaginella* species (*Selaginella tamariscina* (Beauv.) Spring, *Selaginella nipponica*, and *Selaginella pachystachys*)<sup>15</sup>. From then on, phytochemical researchers have isolated and identified this biflavonoid from more than 120 plants, some of which have been used as traditional folk medicines in many regions of the world for even thousands of years. With the development of modern pharmacology, more and more evidence has proved many of the bioactivities of amentoflavone, including anti-oxidant<sup>16</sup>, anti-inflammatory

17, anti-senescence<sup>[18]</sup>, anti-tumor<sup>19</sup>, anti-virus<sup>20</sup>, and anti-fungal<sup>21</sup> effects, as well as therapeutic effects on the central nervous system<sup>22</sup> and cardiovascular system<sup>23</sup>, etc. The purpose of this study is to identify and characterize the bioactive principles from the whole plant of *S. bryopteris*. In this paper, we report the isolation and characterization of known compounds from *S. bryopteris* namely amentoflavone.

## MATERIALS AND METHODS

### Plant materials

Whole plant of *S. bryopteris* was collected from Village Rainikheda and Tamiya, Dist. Chhindwara, (M.P.) India. Herbarium of plants species were prepared graciously and submitted to Department of Botany, Saifia College of Science, Bhopal India, for authentication. Plants were authenticated by Dr. Zia-Ul-Hasan, Head, Department of Botany, Safia College of Science, Bhopal, India. Plant authentication voucher numbers obtained were 391/Bot/Saifia/16 for *S. bryopteris*.

### Chemical reagents

All the chemicals used in this study were obtained from Hi Media Laboratories Pvt. Ltd. (Mumbai, India), Sigma Aldrich Chemical Co. (Milwaukee, WI, USA), SD Fine-Chem. Ltd. (Mumbai, India) and SRL Pvt. Ltd. (Mumbai, India). All the chemicals used in this study were of analytical grade.

### Extraction

Collected plant material washed under running tap water and kept in shade for drying. Dried plant materials were then powdered using blender and further observed for colour, odour, and texture then placed in packed labelled air tight container for further use. Plant material was extracted by continuous hot percolation method using Soxhlet apparatus<sup>24</sup>. Powdered material of *S. bryopteris* was placed in thimble of soxhlet apparatus. Soxhlation was performed at 60°C using petroleum ether as non-polar solvent. Exhausted plant material (marc) was dried and afterward re-extracted with ethyl acetate and methanol solvent. For each solvent, soxhlation was continued till no visual colour change was observed in siphon tube and completion of extraction was confirmed by absence of any residual solvent, when evaporated. Obtained extracts was evaporated using rotary vacuum evaporator (Buchi type) at 40°C. Dried extract was weighed and percentage yield for each extract was determined.

### Qualitative phytochemical analysis of plant extract

The *S. bryopteris* whole plant extract obtained was subjected to the preliminary phytochemical analysis following standard methods<sup>25,26</sup>. The extract was screened to identify the presence or absence of various active principles like phenolic compounds, carbohydrates, flavonoids, glycosides, saponins, alkaloids, fats or fixed oils, protein and amino acid and tannins.

### Quantification of secondary metabolites

Quantitative analysis is an important tool for the determination of quantity of phytoconstituents present in plant extracts. For this TPC and TFC are determined. Extracts obtained from whole plant of *S. bryopteris* plant material of subjected to estimate the presence of TPC and TFC by standard procedure.

### Total phenolic content estimation

The amount of total phenolic in extracts was determined with the Folin Ciocalteu reagent. Concentration of (20-

100µg/ml) of gallic acid was prepared in methanol. Concentration of 100 µg/ml of plant extract were also prepared in methanol and 0.5ml of each sample were introduced in to test and mixed with 2 ml of a 10 fold dilute folin Ciocalteu reagent and 4 ml of 7.5% sodium carbonate. The tubes were covered with parafilm and it was then incubated at room temperature for 30 min with intermittent shaking and the absorbance were taken at 765 nm against using methanol as blank. Total phenolic content was calculated by the standard regression curve of gallic acid and the results were expressed as gallic acid equivalent (mg/g)<sup>27</sup>.

### Total flavonoid content estimation

Different concentration of rutin (20 to 100µg/ml) was prepared in methanol. Test sample of near about same polarity (100µg/ml) were prepared. An aliquot 0.5ml of diluted sample was mixed with 2 ml of distilled water and subsequently with 0.15 ml of a 5% NaNO<sub>2</sub> solution. After 6 min, 0.15 ml of a 10% AlCl<sub>3</sub> solution was added and allowed to stand for 5min, and then 2 ml of 4% NaOH solution was added to the mixture. The final volume was adjusted to 5ml with distilled water and allowed to stand for another 15 min. Absorbance was determined at 510 nm against water as blank. Total flavonoid content was calculated by the Standard regression curve of Rutin/ Quercetin<sup>27</sup>.

### Isolation of sitosterol by preparative TLC

#### Preparation of stationary phase

Readymade silica gel GF 254 plates with a layer thickness of 0.25 mm, dimension 20 cm×20 cm. The plates were reactivated by heating in the oven at 100°C for 15 min, left to cool, and used for application after allocation of the baseline and the solvent front.

#### Preparation of solvent system

Mobile phase for biflavonoid, chloroform-methanol 75:25 (v/v) was mixed in a conical flask and introduced in the jar. The jar was lined with a filter paper, closed tightly, and left for saturation.

### Spectroscopic characterization

Different spectroscopic methods were used to elucidate the structure of isolated compound. Among the spectroscopic techniques IR, <sup>1</sup>H-NMR and MASS were carried out. The IR spectrum was recorded on FTIR Perkin Elmer, <sup>1</sup>H-NMR and spectra were recorded using CDCl<sub>3</sub> as solvent on Bruker Advance II 400 NMR spectrometer from RGPV, University, Bhopal mass spectra were recorded at high resolution on a mass spectrometer (Perkin Elmer Auto system) at Sophisticated Instrumentation centre for Indian Institute of Science Education and Research (IISER) Bhopal, Madhya Pradesh, India and the data are given in m/z values.

## RESULTS AND DISCUSSIONS

The crude extracts so obtained after each of the successive soxhlation extraction process were concentrated on water bath by evaporation the solvents completely to obtain the actual yield of extraction. The yield of extracts obtained from the whole plants of the *S. bryopteris* using petroleum ether, ethyl acetate and methanol as solvents are depicted in the Table 1. The petroleum ether, ethyl acetate and methanol extract of *S. bryopteris* was subjected to screening for its phytochemical constituents. The phytochemical screening results are shown in Table 2. The methanolic extracts containing alkaloids, carbohydrates, reducing sugar, flavonoids, protein, amino acid, tannin and phenolic.

Quantitative phytochemical assay was performed by calculating total phenolic content (TPC) and total flavonoid content (TFC). The TPC was calculated with respect to gallic acid (standard) and TFC was then calculated with respect to rutin taken as standard. The TPC and TFC in methanolic extract were found to be 184.13±0.416mg/gm and 89.67±2.516mg/gm respectively table 3 & fig 1, 2.

During extraction, solvents methanol extract of the plant was investigated by TLC which revealed the presence of amentoflavone that appeared as spot in mobile phase (chloroform: methanol) against amentoflavone reference standard, and the spot of extract appeared the same R<sub>f</sub> value as that in reference standard on TLC plate as shown in Table 4, as indicated by the development of violet spots after

spraying by vanillin-sulfuric acid spray reagent [13]. The IR (KBr) absorption spectrum (Fig. 3) showed absorption peaks:  $\nu$  3244, 3188, 3148 (Broad peak of OH), 1685 (Aromatic C=C), 1604, 1556, 1508  $\text{cm}^{-1}$   $^1\text{H}$ NMR (Fig.4)  $^1\text{H}$ NMR (500MHz,  $\text{CD}_3\text{OD}$ ):  $\delta$  8.80 (d, J = 5.5, 1H), 7.83 (m, 2H), 7.10 (m, 2H), 6.88 (s, 1H), 6.38 (d, J=4.5, 1H) 6.03 (d, J=10, 1H), 4.91 (s, 1H). HRMS  $m/z$ =538 M<sup>+</sup> (Fig. 5).

**Table 1: Results of percentage yield of extracts**

Plant Name	Percentage yield (%)		
	Pet. Ether	Ethyl acetate	Methanol
<i>S. bryopteris</i>	1.48	2.67	2.49

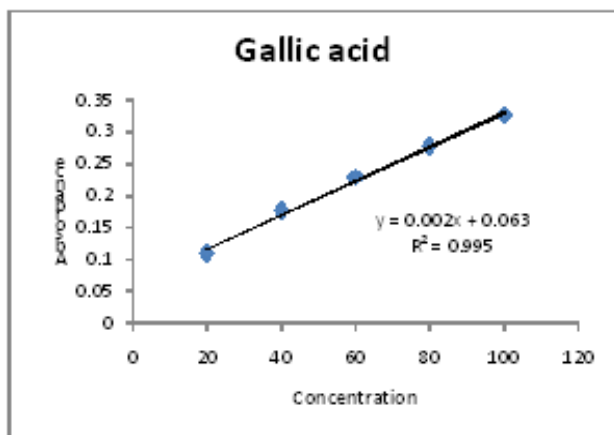
**Table 2: Phytochemical evaluation of *S. bryopteris* extract**

S. No.	Experiment	Result (SB) Extract		
		Pet. ether	Ethyl acetate	Methanolic
<b>1. Alkaloids</b>				
1.1	Mayer's reagent test	-ve	-ve	+ve
1.2	Wagner's reagent test	-ve	-ve	+ve
1.3	Hager's reagent test	-ve	-ve	+ve
<b>2. Carbohydrates</b>				
2.1	Molish's test	-ve	+ve	+ve
2.2	Barfoed's test	-ve	+ve	+ve
<b>3. Test for Reducing Sugar's</b>				
3.1	Fehling's test	-ve	-ve	+ve
3.2	Benedict's test	-ve	-ve	+ve
<b>4. Flavonoids</b>				
4.1	Alkaline reagent test	-ve	+ve	+ve
4.2	Shinoda test	-ve	+ve	+ve
4.3	Lead acetate test	-ve	+ve	+ve
<b>5. Glycoside</b>				
5.1	Borntrager test	-ve	+ve	-ve
5.2	Legal's test	-ve	+ve	-ve
5.3	Killer- Killiani test	-ve	+ve	-ve
<b>6. Tannin and Phenolic compound</b>				
6.1	Ferric chloride test	-ve	+ve	+ve
6.2	Lead Acetate test	-ve	+ve	+ve
6.3	Dilute Iodine solution	-ve	+ve	+ve
<b>7. Saponin</b>				
7.1	Faom Test	-ve	-ve	-ve
<b>8. Test for Proteins and amino acid</b>				
8.1	Ninhydrin test	-ve	-ve	+ve
<b>9. Test for Triterpenoids and Steroids</b>				
9.1	Salwonski Test	-ve	-ve	-ve
9.2	Libberman-Burchard's	-ve	-ve	-ve

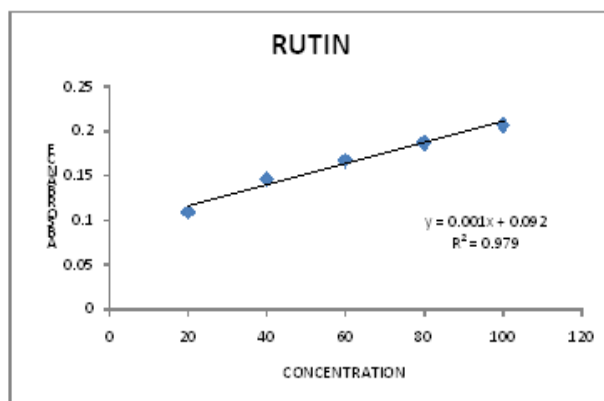
+ve: Present; -ve: Absent

**Table 3: Total phenolic and flavonoid content of extracts**

Test	Methanolic extract
TPC	184.13±0.416mg/gm equivalent to Gallic acid
TFC	89.67±2.516mg/gm equivalent to Rutin



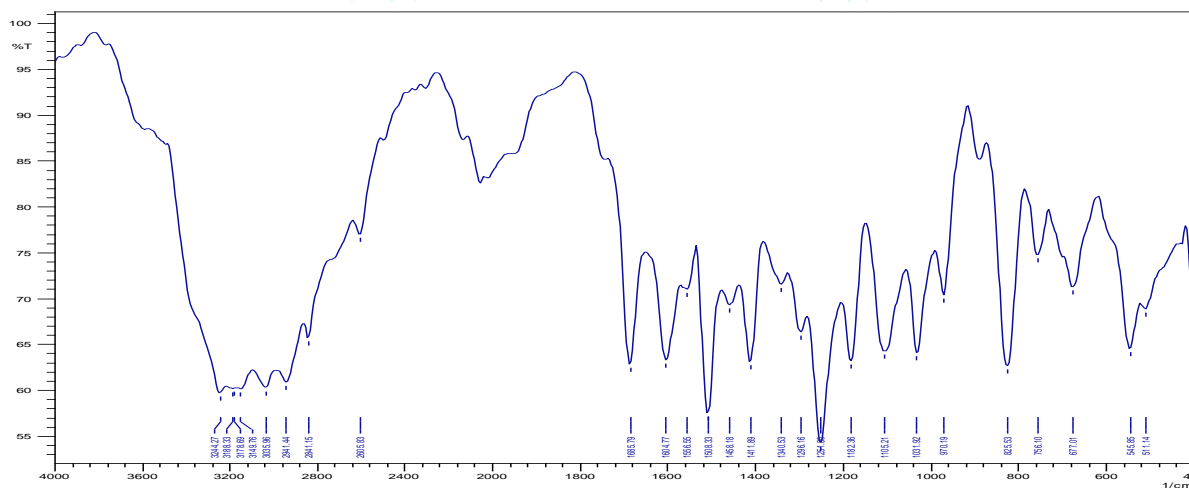
**Figure 1: Graph of estimation of total phenolic content**



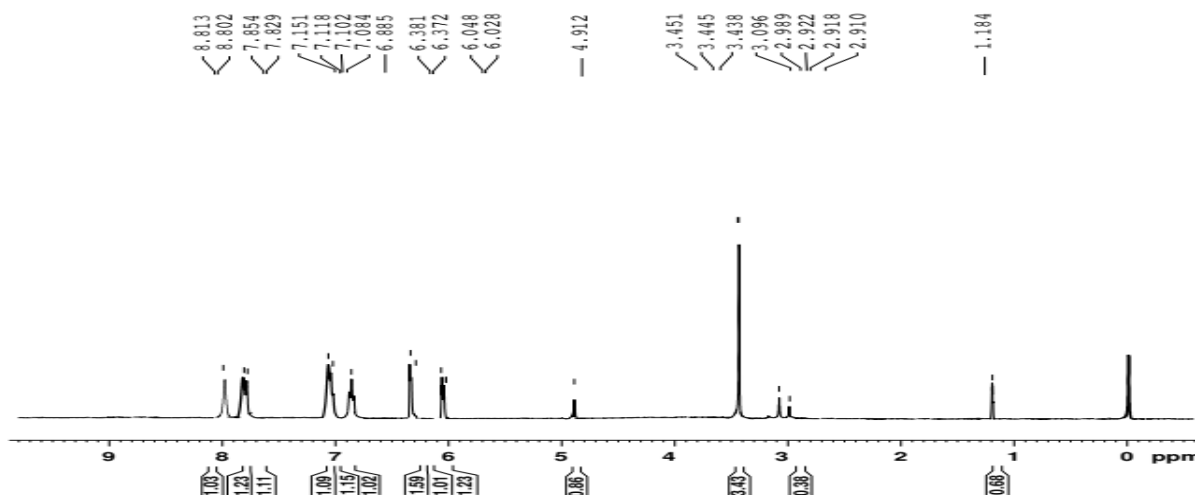
**Figure 2: Graph of estimation of total flavonoids content**

**Table 4: Rf value of standard and extract**

Solvent system	Chloroform: methanol
Rf value of amentoflavone standard	0.232
Rf value of amentoflavone in extract	0.234



**Fig. 3: Fourier transforms infrared spectra of the amentoflavone**



**Fig. 4: <sup>1</sup>H NMR Spectrum of the amentoflavone**

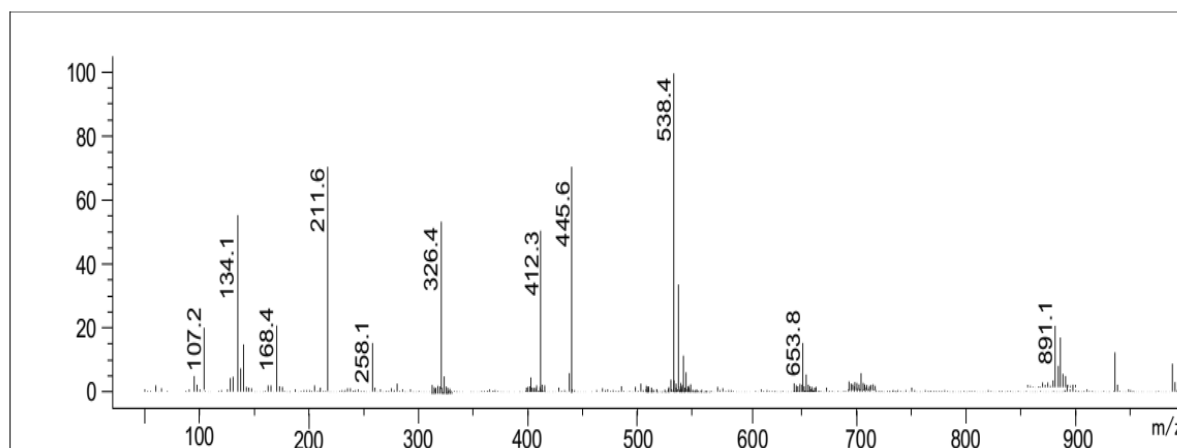


Fig. 5: Mass spectrum of the amentoflavone

## CONCLUSION

From the above findings, amentoflavone were isolated from methanolic extract of the whole plant of *S. bryopteris* and chemical structures elucidated respectively. It was carried out by means of various physical (solvent extraction, TLC, Column chromatography) and spectral techniques.

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