Antioxidant activity, Total Phenolic and Flavonoid Contents and Cytotoxic activity of Euphorbia aegyptiaca

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ABSTRACT

Euphorbia aegyptiaca is a herbaceous plant traditionally used in Sudan for treatment of various diseases, and the study of this plant is still limited. The aim of the present study was to screen the phytochemicals and to assess the Antioxidant activity, total phenolic, flavonoid contents and cytotoxic activity of Euphorbia aegyptiaca. The plant material was extracted successively by Soxhlet apparatus using n-hexane, chloroform and methanol. The chemical constituents of the extracts were carried out using the standard procedures. The Folin-Ciocalteu and Aluminium chloride method was employed to calculate the total phenolic and flavonoid content, respectively. The antioxidant activity, was assessed by measuring the scavenging activity of the DPPH (2,2-Di (4-tert-octylphenyl)-1-picryl-hydrazyl) and Propyl Gallate as standard antioxidants. While cytotoxic activities were screened using brine shrimp. Phytochemical screening studies revealed that flavonoids, tannins, coumarins, saponins, sterols, terpenes, anthraquinones and alkaloids were the main phytochemicals present in extracts of E. aegyptiaca. The methanol extract showed the highest level of total phenolic contents (173.49±2.427 mg GAE/g) and flavonoid content (239.53±7.90 mg QE/g), and the highest antioxidant activity 89% with least (IC50 0.0449µg/ml), and the no toxicity against brine shrimp (LD50 3423.156). Furthermore, no toxicity in all extracts was observed. The present study is the first evaluation regarding the characterization of E. aegyptiaca and its safety, and the results demonstrate its antioxidant potential and suggest its safe therapeutic use. The results suggest that methanol extract is a rich source of phytochemicals and exhibits highest amount of total phenolic, flavonoid content and significant antioxidant activity and it has no cytotoxic activity. E. aegyptiaca plant can be regarded as a promising Source of naturally occurring potential antioxidants.

Keywords: Euphorbia aegyptiaca, Sudan, Antioxidant, Total phenolic, Total flavonoid, Cytotoxicity.

INTRODUCTION

Researches in the field of plant antioxidant focus on antioxidant agents that can protect biological components from oxidative damage [1]. Dietary intake of phenolic compounds correlates with reduced coronary heart disease, cancer mortality and protective in many health-related properties, such as antioxidant, anticancer, antiviral and anti-inflammatory activities [2, 3, 4]. On the other hand, concern about the safety of the commonly used synthetic antioxidants such as butylated hydroxyanisole (BHA) and tertiary butylhydroquinone (TBHQ) has led to increased interest in plant antioxidants [5].

The genus Euphorbia aegyptiaca (Euphorbiaceae) is considered as a large family of flowering plants comprising around 300 genera with 7,500 species [6]. E. aegyptiaca known locally in Sudan as umm lebaina or malhaine. The whole plant is used against scorpion bites. In Sudanes traditional medicine as wellas for treatment of inflammatory conditions like rheumatoid arthritis, conjunctivitis and dermatitis [7]. Euphorbia species contain phytochemical constituents like flavonoids, coumarins, triterpenoids, lignans and alkaloids [8, 9].

To date, to the best of our knowledge there has been no report on the quantitative determination of total phenolic and flavonoid contents and cytotoxic activity of E. aegyptiaca. The aim of the present study was to screen the phytochemicals and to assess the antioxidant activity; total phenolic and flavonoid content and cytotoxic activity of E. aegyptiaca.
MATERIAL AND METHODS

Sample collection

Whole plant of w as collected in the White Nile state, Sudan, during February 2016. The identity of the plant was confirmed at the Medicinal and Aromatic Plants Research Institute (MAPRI), Khartoum, Sudan and the voucher specimens were deposited in the herbarium. The plant was dried for 15 days under the shade, then pulverized by mechanical grinder and stored in air-tight containers till use.

Chemicals

The chemicals and reagents used were of analytical grade. n-Hexane, Chloroform, Methanol, Dimethyl sulfoxide (DMSO), Gallic acid, ascorbic acid, aluminum chloride, sodium acetate and sodium carbonate were purchased from SDFL, India. Ethanol LR, was obtained from Duksan, South Korea. 2.2Dpi (4-tert-octylphenyl)-1-picyrly-hydrazyl (DPPH) and Quercetin were purchased from Sigma-Aldrich, UK. Foline-Ciocalteu reagent was purchased from CDPH, India.

Apparatus

Grinder (Disk Mill Model, FFC-15, China), Sensitive Balance (Rad wag, As 220/c/2, Poland), Soxhlet apparatus, heat grinder and stored in tight containers under the shade, then pulverized by rotary evaporator (Heidolph OB2000 Heidolph VV2000, Germany), Ultraviolet (UV/vis-s-240 (PC) S, Double beam Spectrophotometer (Shimadzu Japan).

Preparation of extracts

Five hundred grams of plant extracts were used for the tests according to methods described by Harbone [10], Sofowora [11], Martinez & Valencia [12] with few modifications.

Antioxidant activity:

DPPH radical scavenging assay:

The DPPH radical scavenging assay was performed according to the method of Shimada et al., (1992) [13], with some modification. In 96-wells plate, the test samples were allowed to react with 2, 2-Di (4-tert-octylphenyl)-1-picyrly-hydrazylstable free radical (DPPH) for half an hour at 37°C. The concentration of DPPH was kept as (300μM). The test samples were dissolved in DMSO while DPPH was prepared in ethanol. After incubation, decrease in absorbance was measured at λ: 517nm using multiplicate reader spectrophotometer. Percentage radical scavenging activity of samples was determined in comparison with a DMSO treated control group. All tests and analysis were run in triplicate.

IC₅₀ Calculations:

The IC₅₀ test material, which possesses 50% inhibition of free radicals of all the extracts and their fractions, was determined by monitoring the effect of different concentrations ranging from 0.5-0.0035mg/ml. The IC₅₀ of the extracts and their fractions were calculated using EZ-Fit Enzyme Kinetic Program (Perrella Scientific Inc, U. S.A).

Determination of total phenolics content

The Folin Ciocalteu reagent was used for analysis of total phenoic content in plant extracts was determined [14]. Briefly, Test sample of each extract was prepared in methanol and the concentration of 1 mg/ml was used in the analysis. 1ml of the extract was mixed with 1 ml of Folin-Ciocalteu reagent. The solution was kept at 25°C for 5-8 min before adding 4 ml of sodium carbonate solution 7.5 % and adjusting the volume to 16 ml with water. After 2 h, the absorbance was measured at 765 nm against the blank. The samples were prepared in triplicate for each analysis and the mean value of absorbance was obtained. Gallic acid was used as standard for the calibration curve. The content of phenols in the extracts was expressed in terms of gallic acid equivalent (mg of GAE/g of extract).

Determination of total flavonoids content

The total flavonoid content in the examined plant extracts was measured with an aluminum chloride colorimetric assay [15]. A test sample of each extract was prepared in methanol and the concentration of 1 mg/ml was used in the analysis. An aliquot 1ml of extract or a standard solution of Quercetin (0.05, 0.1, 0.2, 0.3, 0.4 and 0.5 mg/ml) was added to a 10ml volumetric flask, containing 4ml of distilled water. To the flask was added 0.3ml 5% NaNO₃. After 5min, 0.3ml 10% AlCl₃ was added. At the sixth minute, 2ml 1M NaOH was added and 2.4ml was added H₂O. The solution was mixed well and the absorbance was measured against a prepared reagent blank at 510 nm with an UV-v is spectrophotometer. The data of the total flavonoids content of the extracts were expressed as milligrams of quercetin equivalents (QE) Per 100 grams dry mass (mg of Q/g of extract). All samples were analyzed in duplicates.

Toxicity testing against the brine shrimp:

Storage of Artemia Salina eggs:

Eggs of Artemia Salina were stored at low temperatures (4°C), to remain viable for many years.

Hatching shrimp:

Brine shrimp eggs, Artemia Salina were hatched in artificial seawater prepared by dissolving 38g of sea salt in one liter of distilled water. After 24-72h incubation at room temperature (37 °C), the larvae were attracted to one side of the vessel with a light source and then collected with a pipette. Larvae were separated from the eggs by aliquoting them three times in small beakers containing artificial seawater.

Brine shrimp assay:

Bioactivity of the extract was monitored by the brine shrimp lethality test [16]. Brine shrimp lethality bioassay was carried out to investigate the cytotoxicity of extracts and 50 mg of Artemia Salina (Leach) eggs was added to a hatching chamber containing artificial Sea water (75ml). The hatching chamber was kept under an inflorescent bulb for 48h for the eggs to hatch into shrimp larvae. twenty mg of test extracts of the various plant species was separately dissolved in 2 ml of methanol, then 500, 50, and 5μl of each solution was transferred into vials corresponding to 1000, 100, and 10 μg/ml respectively. Each dosage was tested in triplicate. Ten larvae of A. Salina Leach (taken 48 – 72h after the initiation of hatching) were added to each vial (Figure 4). The final volume of solution in each vial was adjusted to 5ml with Sea water immediately after adding the shrimps. One drop of dimethyl sulphoxide (DMSO) was added to the test and control vials before adding the shrimps to enhance the solubility of test materials. LD₅₀ values were determined at
95% confidence intervals by analyzing the data on a computer loaded with a “Finney Program.” The concentration at which it could kill 50% larvae (LD$_{50}$) was determined. LD$_{50}$ values below 200ppm are generally considered as significant [17, 18, and 19].

**Statistical analysis:**
Data were presented as means ± S.D. Statistical analysis of assay results were performed using the Microsoft Excel program 2013 and Finney Program.

**RESULTS AND DISCUSSION**

**Percentage Yield of Extract**
The plant material was extracted successively by the Soxhlet extraction method using n-hexane, chloroform and methanol. The results showed the highest yield of methanol (9.461%), followed by chloroform (1.902%) and then n-hexane (1.433%) as seen in Figure 1.

**Chemical Constituents:**
The phytochemical screening of three successive extracts prepared with n-hexane, Chloroform and Methanol, of *E. aegyptiaca* plant revealed the presence of various secondary metabolites: tannins, flavonoids, coumarins, saponins, sterols, terpenes and alkaloids. Anthraquinones and Cyanogenic glycoside were not detected. Phytochemical constituents of the *E. aegyptiaca* plant have been depicted in Table 1.

![Figure 1: Percentage yield of different extracts of *E. aegyptiaca* plant.](image)

<table>
<thead>
<tr>
<th>No.</th>
<th>Plant Constituents</th>
<th>Tests/reagents used</th>
<th>Name of Extracts</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td>n-Hexane</td>
</tr>
<tr>
<td>1</td>
<td>Tannins</td>
<td>Ferric chloride test</td>
<td>+</td>
</tr>
<tr>
<td>2</td>
<td>Flavonoids</td>
<td>ALCL$_3$ 1% test</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td></td>
<td>KOH 1% test</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Shinoda’s test</td>
<td>+</td>
</tr>
<tr>
<td>3</td>
<td>Coumarins</td>
<td>NH$_3$ 10%</td>
<td>+</td>
</tr>
<tr>
<td>4</td>
<td>Saponins</td>
<td>Froth Test</td>
<td>-</td>
</tr>
<tr>
<td>5</td>
<td>Sterols</td>
<td>Salkowski’s Test</td>
<td>+</td>
</tr>
<tr>
<td>6</td>
<td>Terpenes</td>
<td>Liebermann Test</td>
<td>-</td>
</tr>
<tr>
<td>7</td>
<td>Anthraquinones</td>
<td>Borntrager’s test</td>
<td>-</td>
</tr>
<tr>
<td>8</td>
<td>Alkaloids</td>
<td>Mayer’s reagent</td>
<td>+</td>
</tr>
<tr>
<td>9</td>
<td>Cyanogenic glycosides</td>
<td>Guignard test</td>
<td>-</td>
</tr>
</tbody>
</table>

Key: + =Trace, ++ =Moderate, +++ =High, - =Negative.

**Total phenolic and total flavonoid contents**
The methanol extract showed the highest level of total phenolic contents (173.49±2.427 mg of GAE/g) and flavonoid content (239.53±7.90 mg QE/g), Table 2. Plants with high phenolics and flavonoids contents possess remarkable antioxidant activity due to their redox properties and chemical structures.

<table>
<thead>
<tr>
<th>NO.</th>
<th>Extract</th>
<th>Total Phenolic content (mg of GAE/g of extract)</th>
<th>Total Flavonoid content (mg of QE/g of extract)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>n-Hexane</td>
<td>15.823±0.348</td>
<td>55.072±4.56</td>
</tr>
<tr>
<td>2</td>
<td>Chloroform</td>
<td>18.50±0.0483</td>
<td>26.087±13.69</td>
</tr>
<tr>
<td>3</td>
<td>Methanol</td>
<td>173.49±2.427</td>
<td>239.53±7.90</td>
</tr>
</tbody>
</table>

Key: GAE= Gallic acid equivalent, QE= Quercetin equivalent, Values are expressed as means of triplicate determinations ± SD.

**CONCLUSION**

It is proved that the methanol extract of *E. aegyptiaca* has the highest antioxidant activity, total phenolic and flavonoid contents. Furthermore, have no toxic effect. Therefore, it can be stated that *E. aegyptiaca* plant can be regarded as a promising candidate of natural antioxidant compounds with high values. These results might be helpful for treating diseases related to oxidative processes. The results of our investigations is a good starting points for Fractionation, isolation, purification and monitoring of activities of constituent according to their antioxidant effect. Identification and characterization of the chemical structure of these phenolics and flavonoids of the methanolic extract of *E. aegyptiaca*, Could be a remarkable addition to the global database.

**REFERENCES**


